

Worksheet: Surface Tension and Soap Films

Name: _____ Group name: _____

In this lab you will study surface tension in water and soapy water and its effects. Please **perform the experiments in the trays provided** to reduce spillage. Leave the bonus assignments for last, if you still have time.

1. Fill a gray plastic cup to the brim with water. Carefully add water to the plastic cup by emptying eye-droppers below the surface of the water.

- Observe the shape of the water surface (“**meniscus**”) as you add more and more water by looking horizontally at the cup. Draw the shape below. How can you explain this shape?

- Add more and more water until the cup overflows. Estimate the height of the meniscus above the cup edge just before it collapses.

Maximum height of the meniscus: _____

- Again start adding water using eye-droppers until a well-visible meniscus forms (but don't let it collapse). Carefully add a single drop of soap. What happens? Why?

- **Bonus:** from the _____ you would expect the meniscus to look like a spherical dome or cap. Why is this not the case here? (Hint: tilt the glass sideways. Why does it spill? What forces do the water molecules feel?)

4. Use the plastic framework kit to build the first two geometric shapes as shown on the next page: a **tetrahedron** and a **cube**. Insert the shape into the beaker of soapy water.

- Describe the surfaces they form.

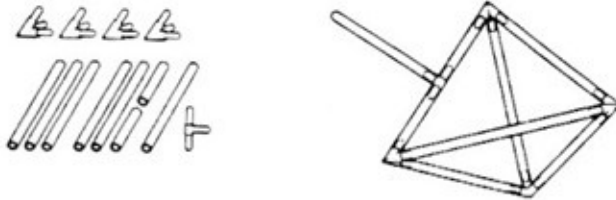
- Puncture one of the film surfaces – how does the shape change?

- What rules or common features exist in all of the films that you have observed (“bubble laws”)?

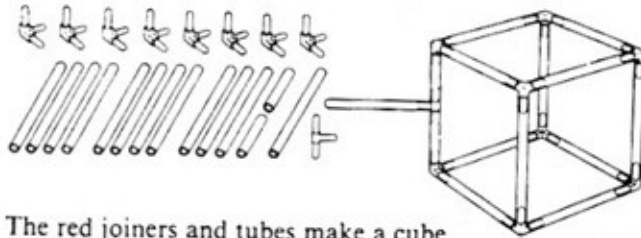
Note: you only need to make the tetrahedron and the cube

Bonus: also study the other two shapes.

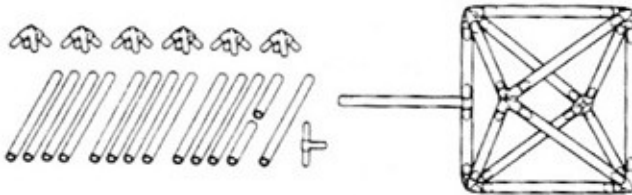
Assembly of the plastic frameworks



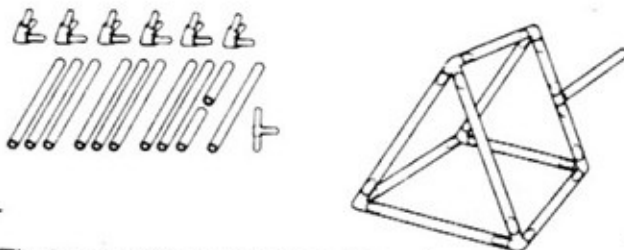
The grey joiners and tubes make a tetrahedron.



The red joiners and tubes make a cube.
White joiner for handle.



The green joiners and tubes make an octahedron.



The blue joiners and tubes make a prism.